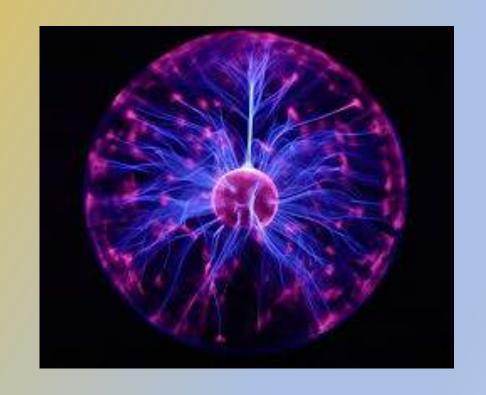
# PLASMA PROCESSES FOR CO<sub>2</sub> RECYCLING

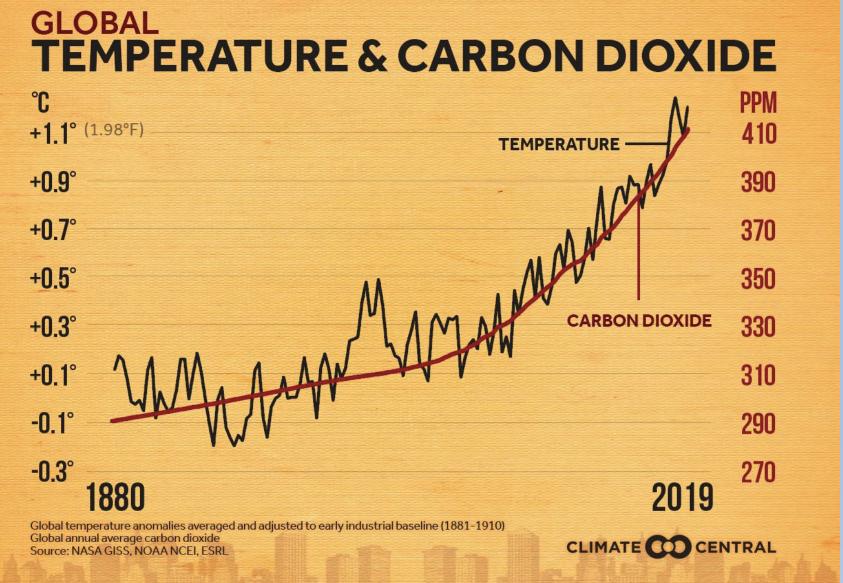


P I John

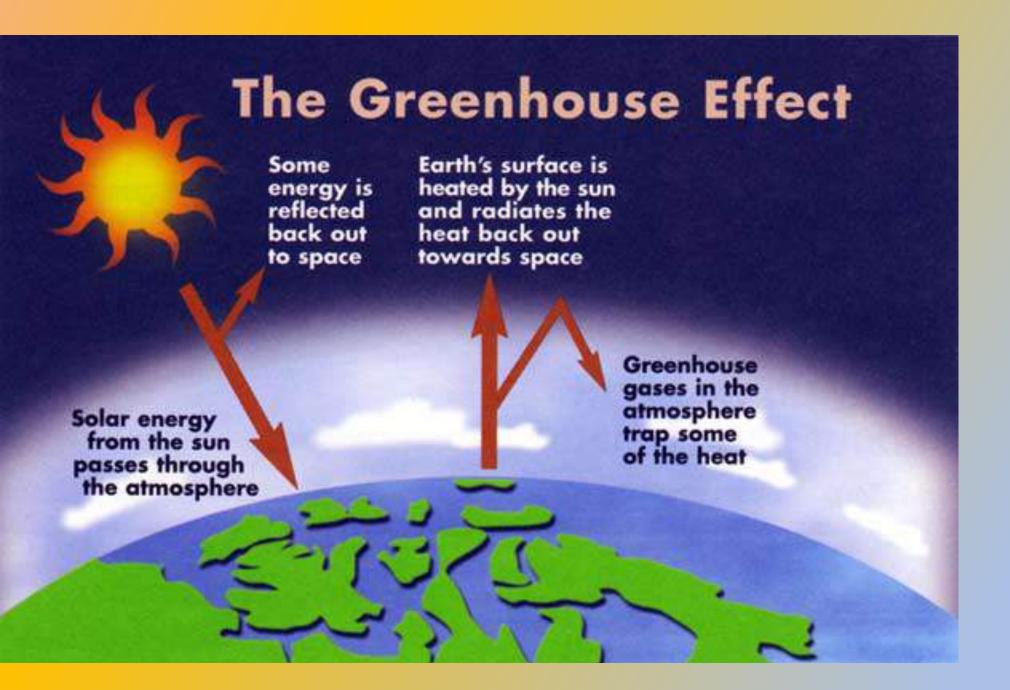
Online Talk given to IIT, Delhi on 4 October 2020



- Anthropogenic CO2 emissions outpace natural carbon cycle
- Atmospheric CO2 increased from 280
   ppm in 1880 to 420 ppm in 2019.
- 3 Trillion tones already accumulated



### **CO2 AND GLOBAL WARMING**



Sunlight falls on earth; scattered back as infrared light

CO2 + GHG layer in the atmosphere absorbs IR and gets heated.

Hot CO2 radiates, increasing atmospheric temperature.

Beyond 1.5 deg C, the risks of extreme climatic events

like floods, drought etc Intergovernmental Panel on Climate Change would be too high



# Global Warming of 1.5°C

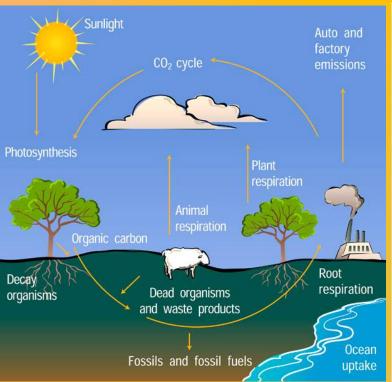
For 1.5 degree rise limit

CO2 emissions should be cut by half by 2030 and brought down to zero by 2055.

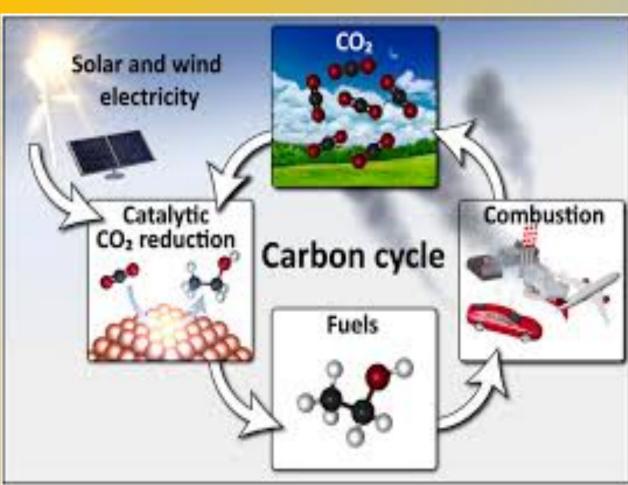
Source: IPCC website www.ipcc.ch

## A NEW SUSTAINABLE CARBON CYCLE

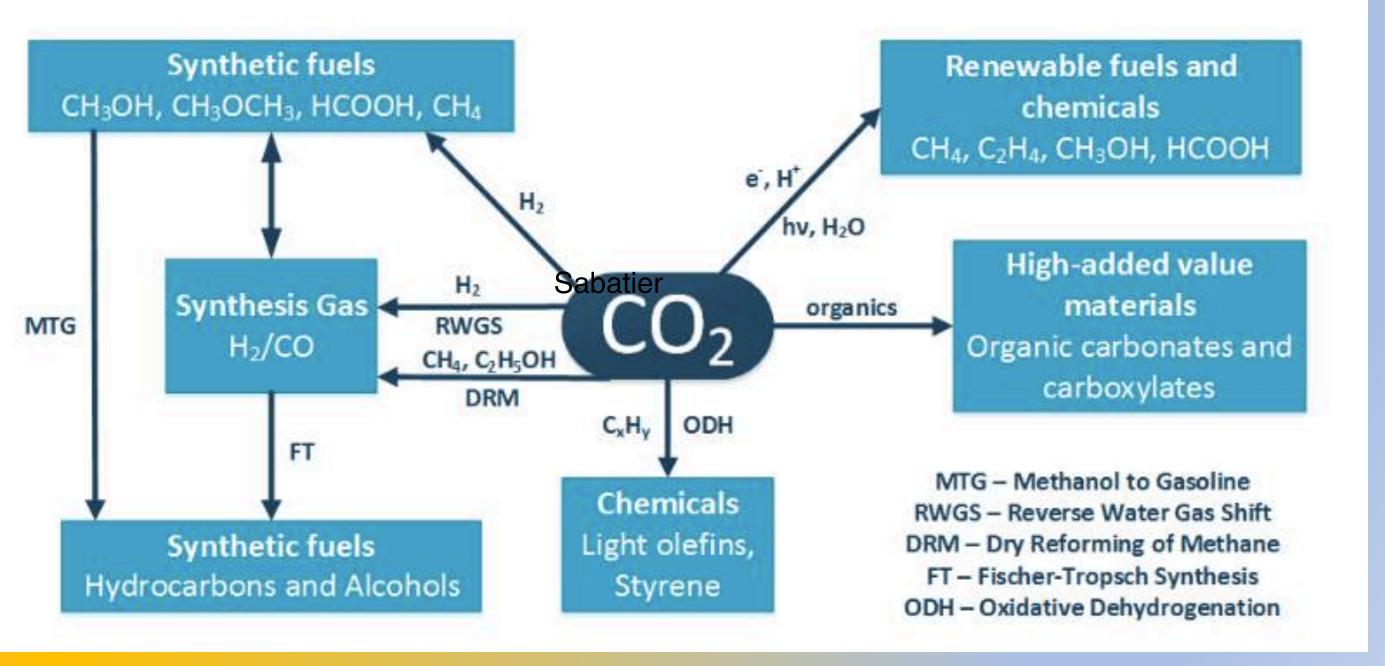




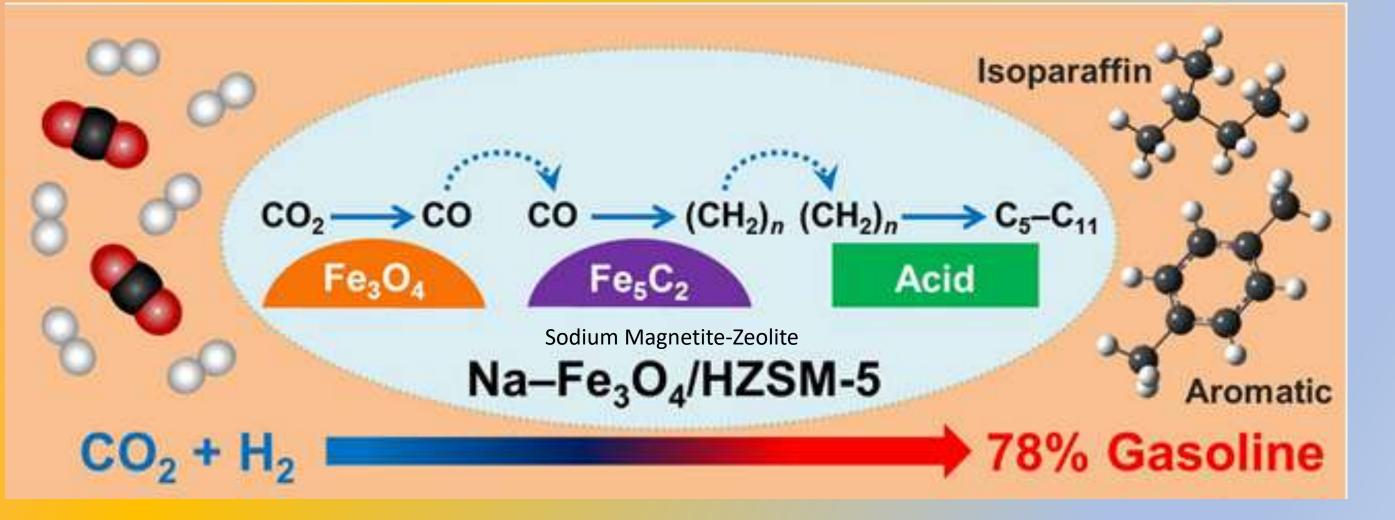
CO2 Recycling by Photosynthesis + Oceans Too slow



- High energy density, compact and safe storage, easy transportation
- Existing hydrocarbon infrastructure can be used
- (Hydrogen, Electric)
- Can transform to net carbon negative system reversing atmospheric CO2 levels



Critical Step: CONVERT CO2 INTO CO



CO2 Methanation (Sabatier Process) 1897 CO2+4H2=CH4+2H2O (-165) Louisa Rui Lin Ting, Oriol Piqué, Si Ying Lim, Mohammad Tanhaei, Federico Calle-Vallejo, and Boon Siang Yeo ACS Catalysis 2020 10 (7), 4059-4069

At 320 deg

New Technologies!

3000 deg For direct thermal splitting

**Catalyst (Ceria)** 

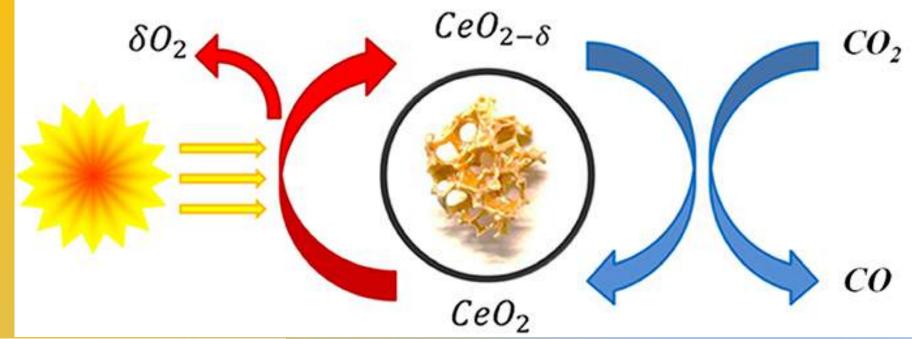
**Solar Thermochemical** 

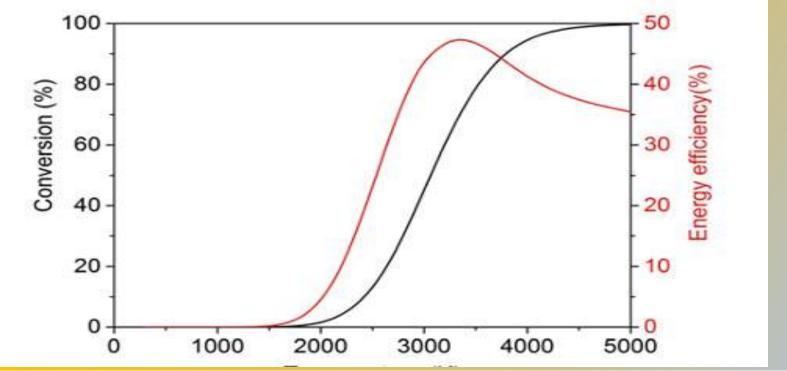


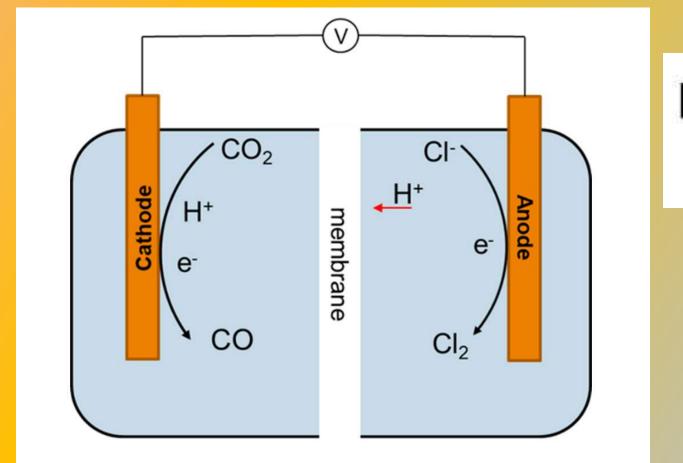
Solar reduction ~1400 °C

2<sup>nd</sup> Step

Non-solar oxidation ~1000 °C







## Electrochemical reaction $CO_2 + 8e^- + 8 H^+ = CH_4 + 2H_2O$

H+ ions from electrolysis
H ions migrate to Cathode through membrane

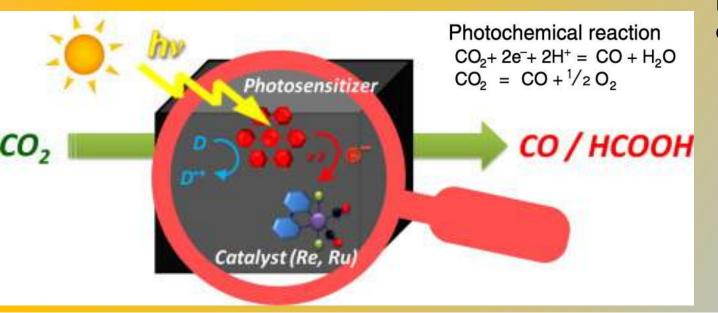


Photo excited electrons in the conduction band of a semiconductor





Most of the matter in the Universe exists as Plasma. Sun, solar wind, stars.

Fluid made of electrons, ions and neutrals, interacting through Coulomb forces and exhibiting collective properties: Waves, Self-organisation





#### **Energetic electrons ionize atoms**

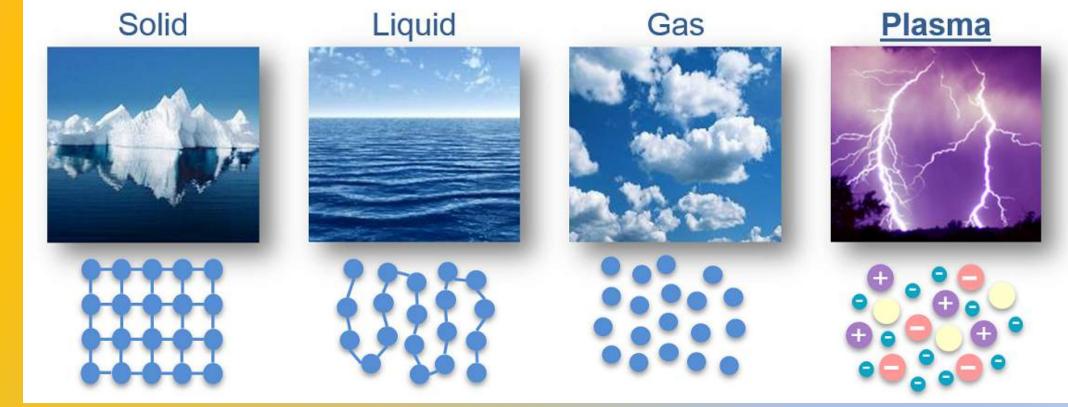
electron-ion pairs.

product electrons gain energy

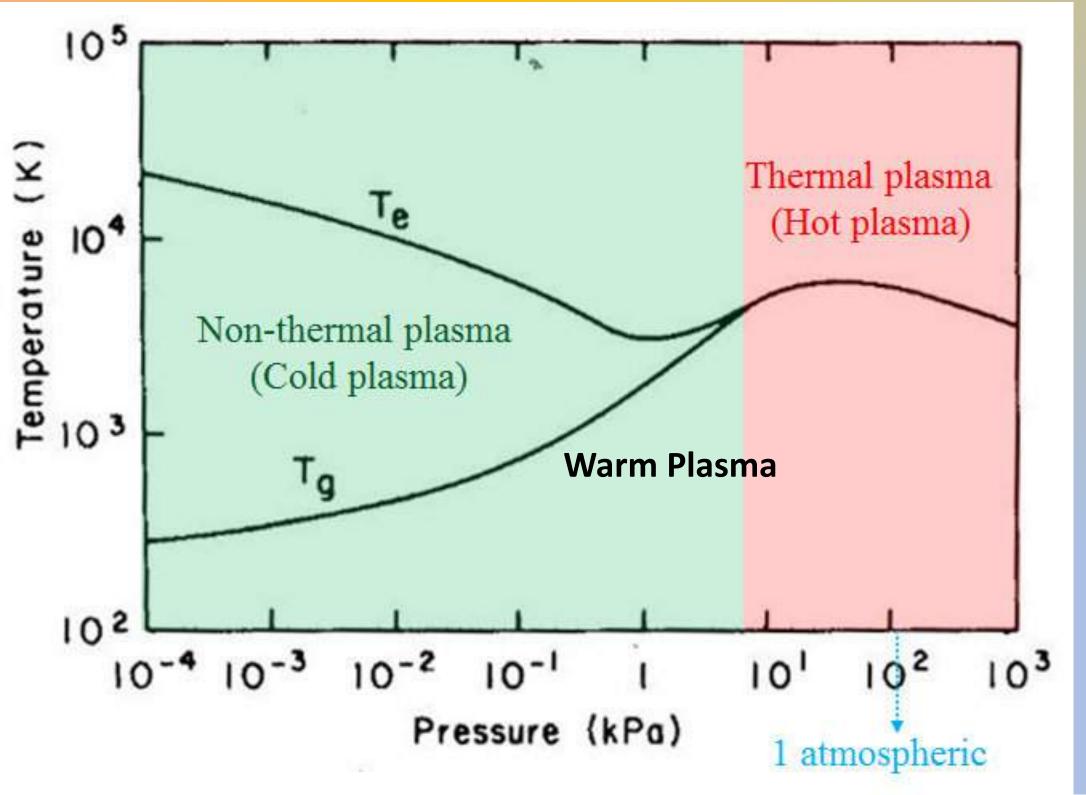
more ionization

**Avalanche of ionization** 

Plasma



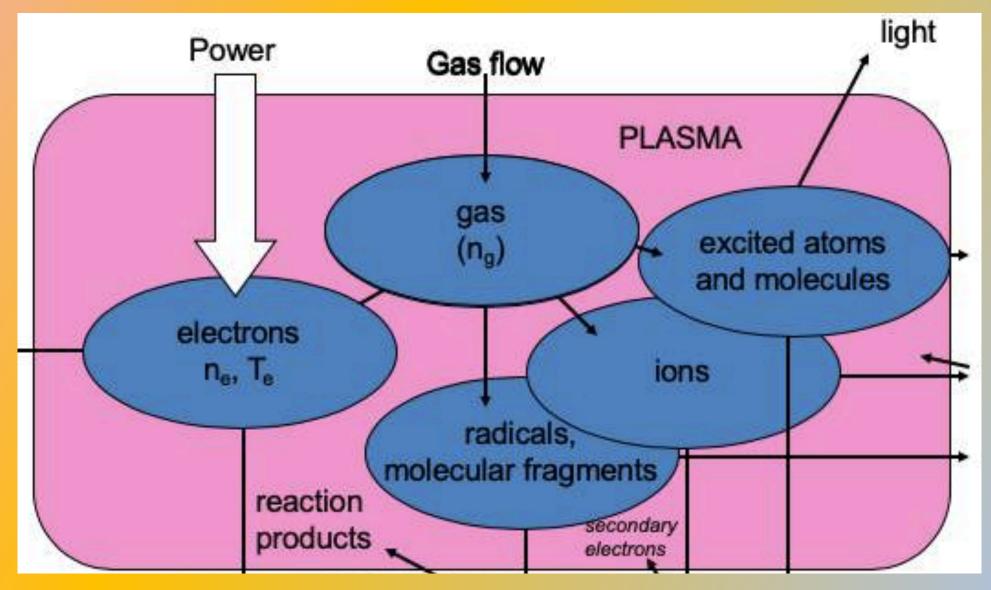




thermodynamically unfavourable chemical reactions

NTP has low density: poor Conversion

Transitional plasma with high Temp and density



High chemical reactivity: free electrons/radicals.

Plasma chemistry

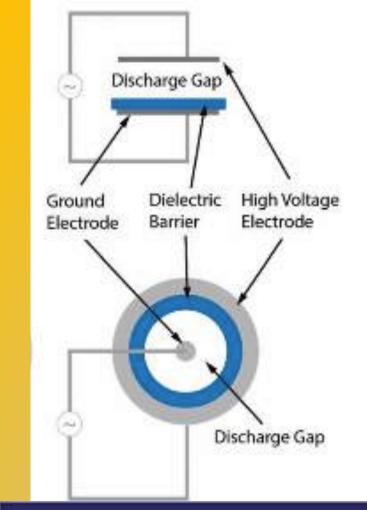
Reactivity depends on plasma parameters like density, temperature, E/p.

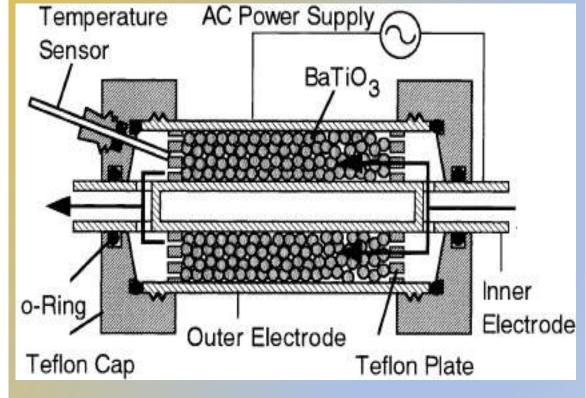
Each plasma source has a unique set of plasma parameters

Many types of plasma sources are required to do process optimization

## DIELECTRIC BARRIER DISCHARGE

- Plane-parallel or concentric metal electrodes
- Dielectric barrier to prevent arcs.
- 'Non-thermal plasma' (streamers)
- Gas is at room temperature
- Electrons at temperatures of 20 000–30 000 K
- Pack with catalysts

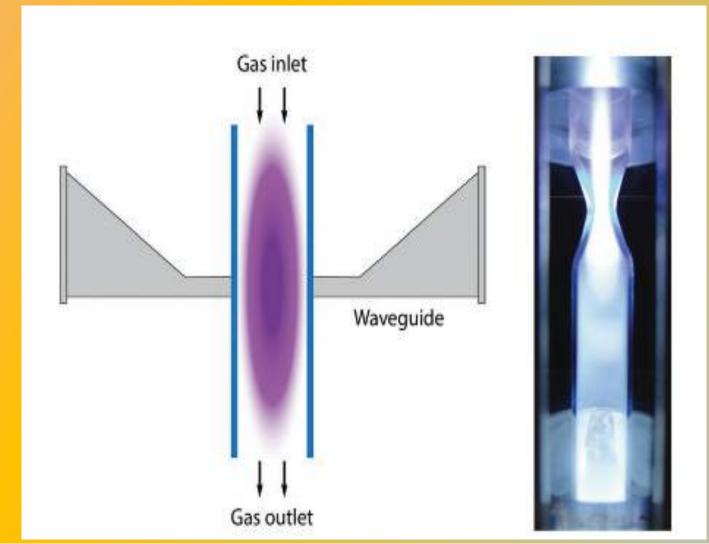




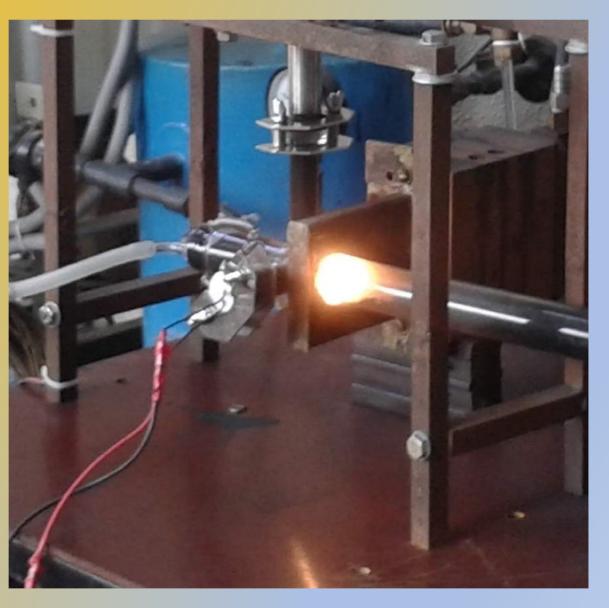


300 MHz to 10 GHz.

Gas flows through a quartz tube.
Intersects with a wave guide (taper)
E-field starts discharge



### MICROWAVE DISCHARGE

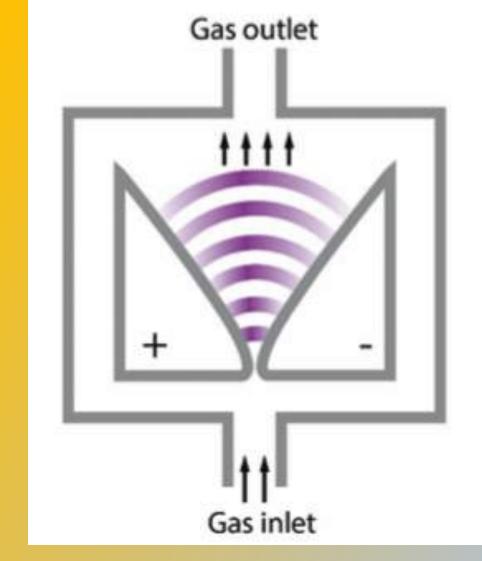


Gas flows between two diverging electrodes with a voltage across it.

Arc plasma formed at the narrow gap

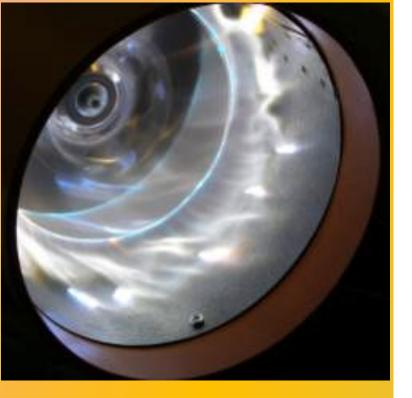
dragged by the gas flow towards a rising interelectrode gap, until it extinguishes.

A new arc is ignited at the shortest interelectrode gap.

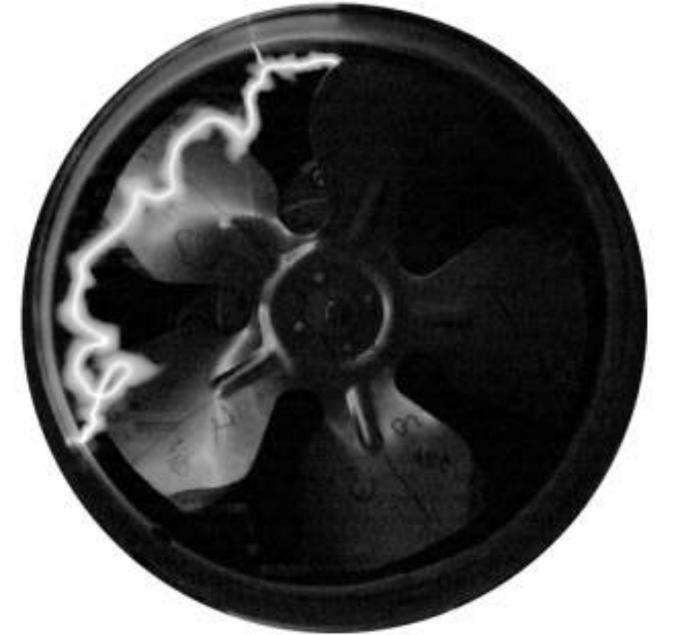




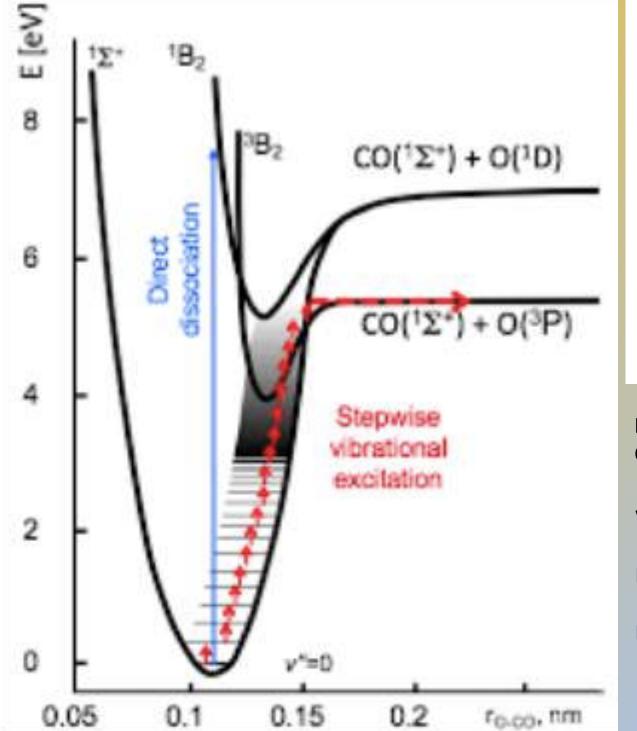
### **GLIDING ARC DISCHARGE**







# Rotating GD Arc Plasma



CO<sub>2</sub>

**Dissociation** 

in Plasma

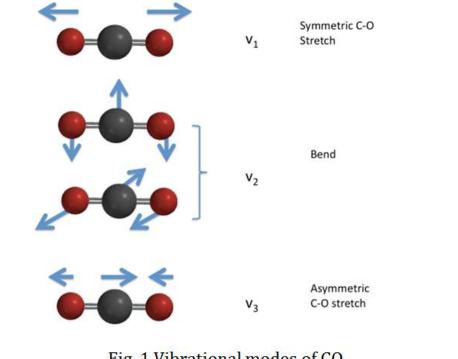


Fig. 1 Vibrational modes of CO<sub>2</sub>

**Electron impact dissociation**  $CO_2 \rightarrow CO + O$ ,  $\Delta H = 5.5 \text{ eV/mol}$ 

Vibrational excitation at 0.29 eV:

**Ladder climbing** 

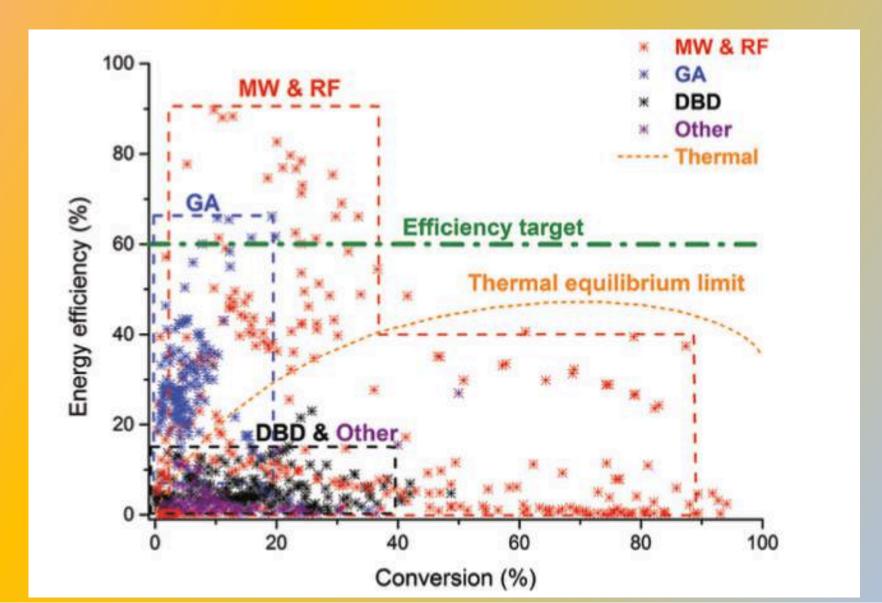
Dissociation

Source: Annemie Bogaerts,

Ramses Snoeckx

Chem. Soc. Rev., 2017, 46, 5805

### PLASMA DISSOCIATION RESULTS



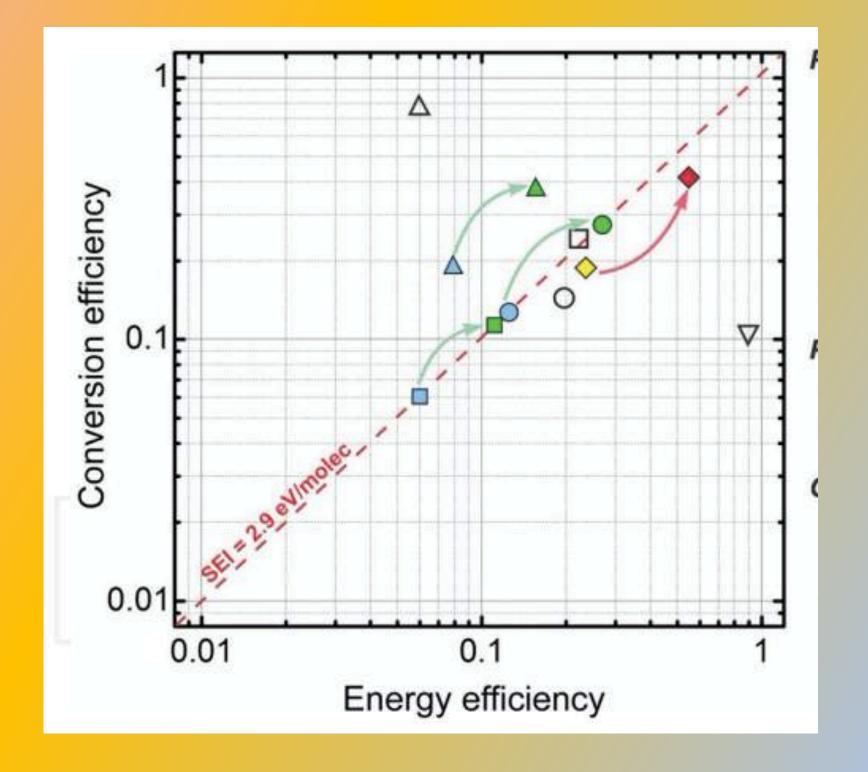
The energy efficiency is a measure of how efficiently the process performs compared to the standard reaction enthalpy

efficiency target of 60% is based on comparison with electrochemical Conversion

MW/GA high performance

**DBD** results

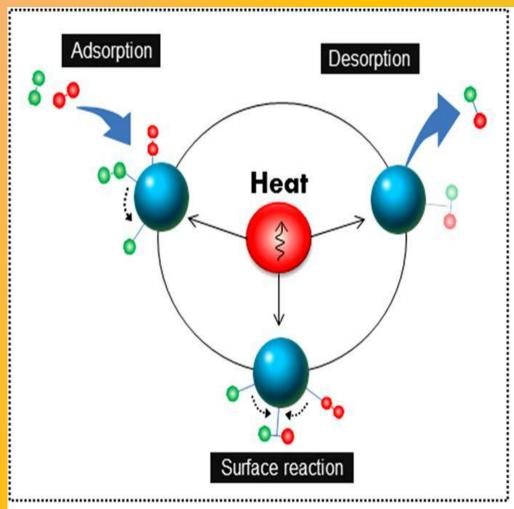
**Low thermal equilibrium results** 



Supersonic gas flow Power Interruption-Red Catalysis-Green

Fine tuning

### **Plasma Catalysis**



Excited molecules Bombardment Plasma Heating Radicals

Vibrationally excited Molecules provide more reactivity at a catalyst surface

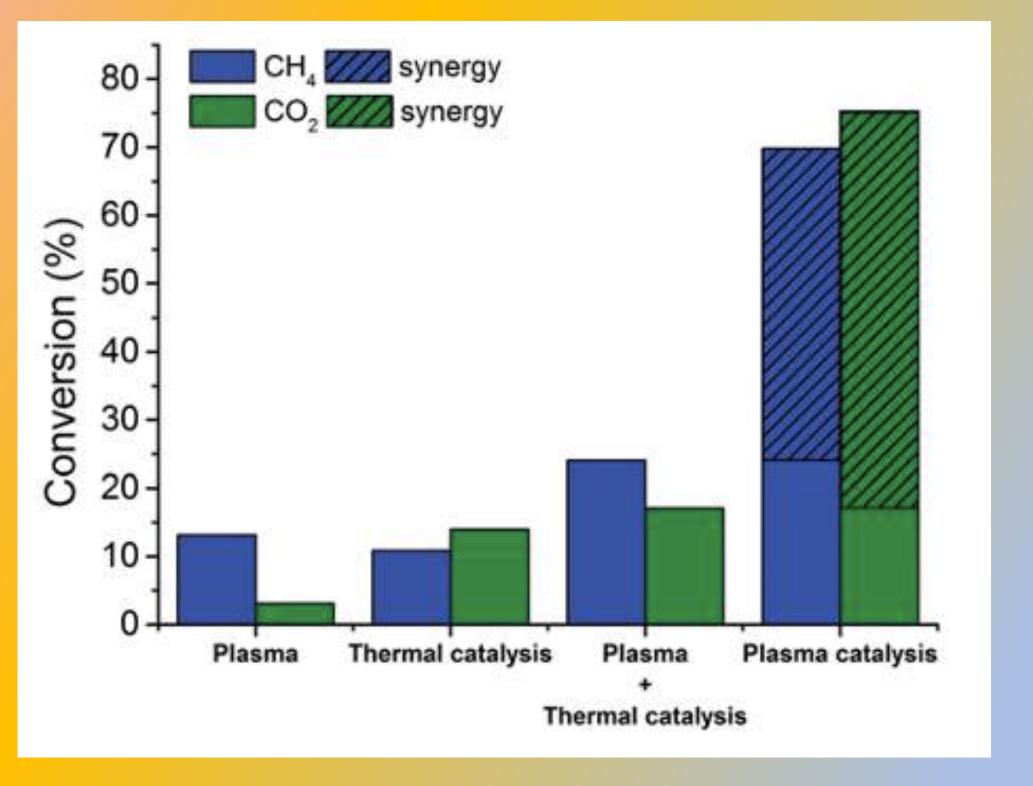
lower barriers for dissociation at the surface.

(a) Thermal catalysis

(b) Plasma-catalysis

Formation of radicals and other excited species Modification of catalyst properties Effect of plasma (physical and chemical) Activation by photon radiation on catalyst Reduction in activation barrier Modification of surface reaction pathways Enhanced energy efficiency Improved selectivity Catalyst Overall Increased catalyst rate and yield **Synergism** synergetic effect Improved catalyst durability Increased concentration of active species Enhanced electric fields Packed bed effect Effect of catalyst Formation of microdischarges in pores on plasma Change in discharge type Pollutant concentration in plasma

Renewable and Sustainable Energy Reviews
Vol 135, Jan 2021



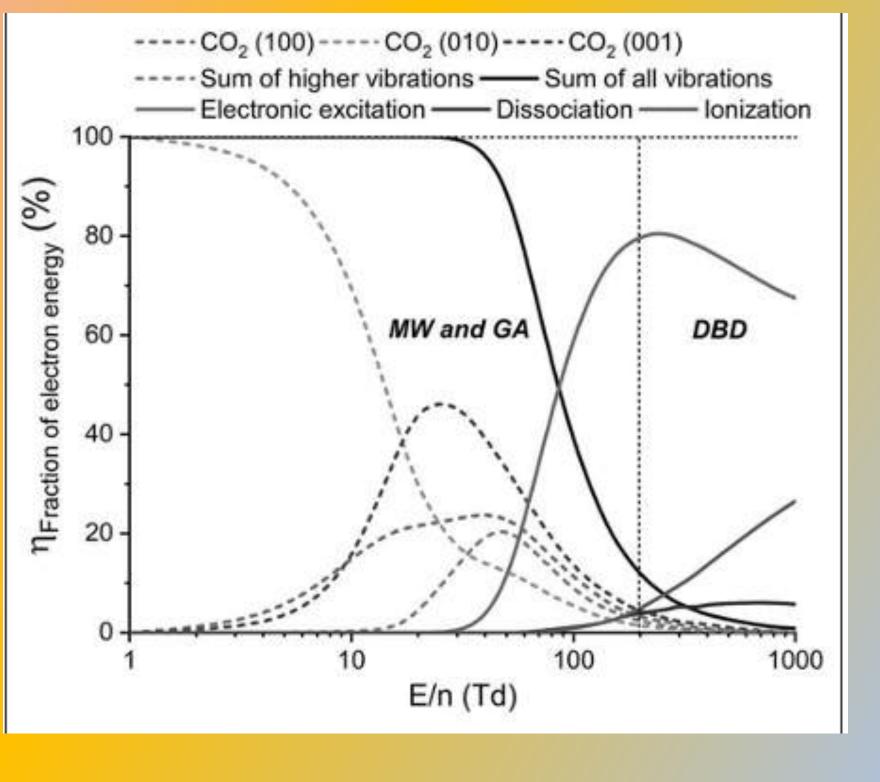
# Plasma Catalysis Synergy

## **Versatility of Plasma Processes**

	CO <sub>2</sub> splitting	CO <sub>2</sub> /CH <sub>4</sub>	CO <sub>2</sub> /H <sub>2</sub>	CO₂/H₂O
Catalysis	X	X	X	NA
Electrochemical	X	NA	NA	X
Solar thermochemical	X	NA	NA	Xª
Photochemical	NA	NA	NA	X
Biochemical	NA	NA	NA	Xp
Plasmachemical	X	X	Xc	X <sup>c</sup>

## PLASMA DISSOCIATION OF CO2: Advantages

- On demand capability
- High energy efficiency (~60% demonstrated)
- High power density (45W/cm³)
- Rapid ramping up and down (wrt high temperature SOEC)
- No scarce materials employed (Pt catalyst in PEM)



## **Designer Plasmas for optimizing conversion**

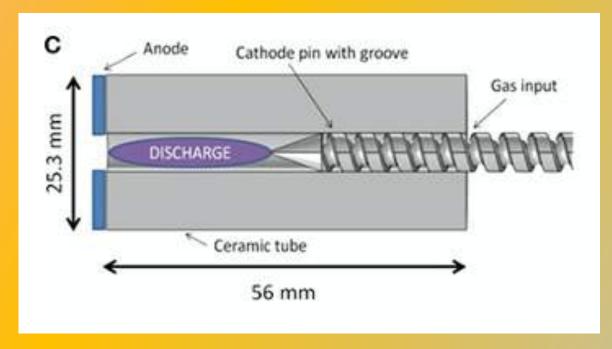
E/n ~ 50Td

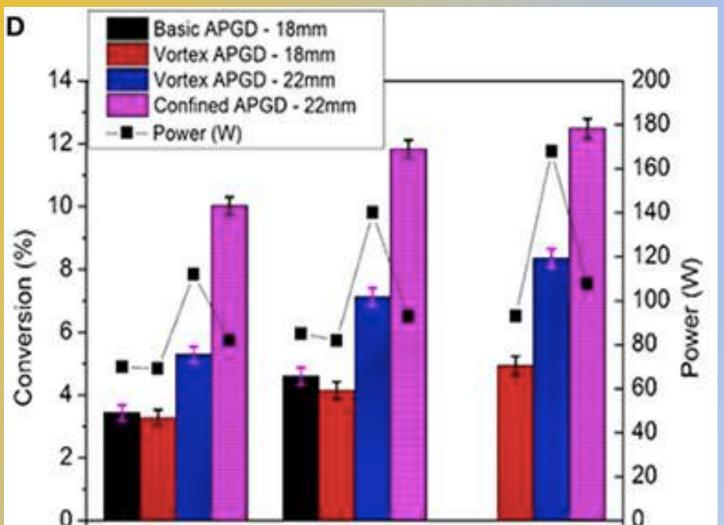
Source: Annemie Bogaerts,

Ramses Snoeckx

Chem. Soc. Rev., 2017, 46, 5805

#### Appropriate plasma devices

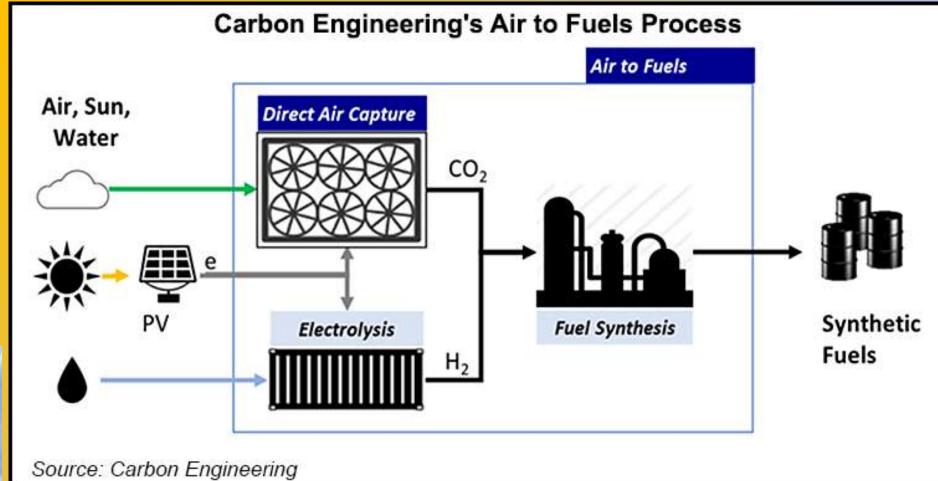




Annemie Bogaerts 1\* and Gabriele Centi, Frontiers Energy Research July 2020









## \$20 MILLION

Prize Purse

The \$20M NRG COSIA Carbon XPRIZE develops breakthrough technologies to convert CO<sub>2</sub> emissions into usable products.









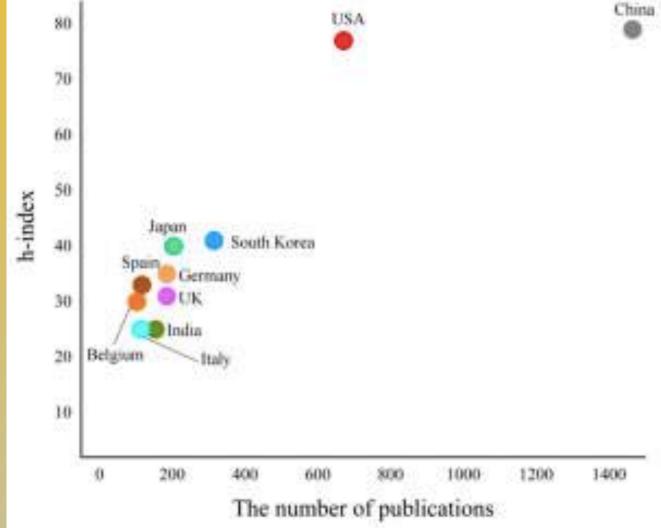








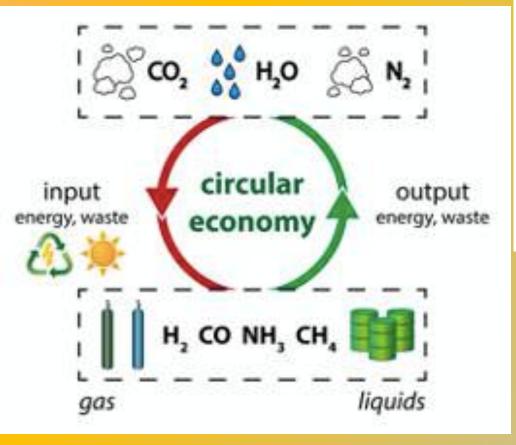


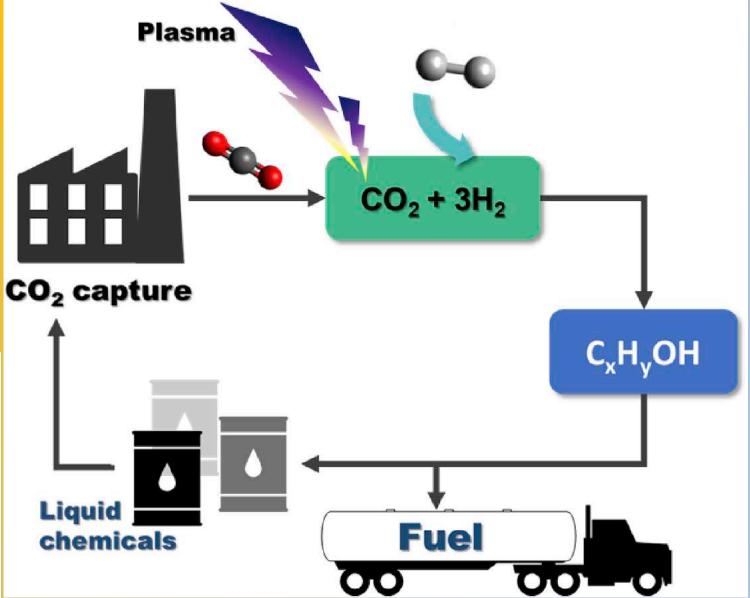


- A new Industrial Revolution based on Circular concepts in waste to value is emerging.
- Europe is the epicentre
- Intellectual fervour and entrepreneurial effervescence.
- Many Startups: including one from India

- CO2 recycling with solar energy
- Storing electricity as chemical energy in gas grids

#### **CO2-Neutral Fuels.**





#### **Summary**

New plasma processes to convert CO2 into fuels enable a sustainable fuel cycle.

It will allow us to continue using the abundant Hydrocarbon resources and existing infrastructure

Plasma Catalysis may further increase the energy efficiency and throughput of the Plasma Process.

More research is needed to improve (i) energy efficiency, (ii) conversion, and (iii) product selectivity (iv) plasma sources

More insight in the plasma-catalyst interactions is crucial for designing catalysts most suited to the plasma environment.



John Pucadyil

## Plasma Processes for Energy and Environment

The Pervasive Role of Plasma Processing in Technologies for Clean Energy and Environment.

Pucady

